

<b>Module Title:</b>	Analytical Techniques Pharmaceutical Science
<b>Language of Instruction:</b>	English
<b>Credits:</b>	10
<b>NFQ Level:</b>	6
<b>Module Delivered In</b>	<a href="#">3 programme(s)</a>
<b>Teaching &amp; Learning Strategies:</b>	This module will be taught in four theory classes of 1 hour duration per week plus 16 x 3 hour labs delivered on a rota. Lectures will be linked to practical classes and the practical component will allow students to develop the required technical competencies, attitudes and behaviours, and promote good record keeping and timely submission of reports.
<b>Module Aim:</b>	The aim of this module is to impart knowledge of basic analytical principles together with practical applications in pharmaceutical and related industries
<b>Learning Outcomes</b>	
<i>On successful completion of this module the learner should be able to:</i>	
LO1	Explain and demonstrate a range of analytical techniques and discuss the theoretical background on which they are based.
LO2	Operate analytical techniques used in the analysis of pharmaceuticals.
LO3	Comprehend and apply quality control measures in pharmaceutical analysis.
LO4	Carry out pharmaceutical assays and chromatographic separations of drugs.
LO5	Identify certain categories of drugs.
<b>Pre-requisite learning</b>	
<b>Module Recommendations</b> <i>This is prior learning (or a practical skill) that is recommended before enrolment in this module.</i>	
No recommendations listed	
<b>Incompatible Modules</b> <i>These are modules which have learning outcomes that are too similar to the learning outcomes of this module.</i>	
No incompatible modules listed	
<b>Co-requisite Modules</b>	
No Co-requisite modules listed	
<b>Requirements</b> <i>This is prior learning (or a practical skill) that is mandatory before enrolment in this module is allowed.</i>	
Successful completion of year 1 or equivalent	

## Module Content & Assessment

### Indicative Content

#### Analytical Techniques and Procedures

General approach to analytical procedures. Steps in analytical analysis. Representative sampling, sample preparation, analysis and interpretation of results. Systematic error and random error. Precision and accuracy. Measurement uncertainty.

#### Titrimetric Methods:

Review of concentration calculations, ppm, w/w,w/v dilutions of solutions. Principles of titrations

#### Spectroscopic Methods

Properties of light. Electromagnetic spectrum. Absorption of light. Beers Law. Absorption Spectra and colour. Spectrometer - good operating techniques. Applications of Beers' Law. Principles of atomic absorption and flame emission spectroscopy, sample treatment, use of standards. Interferences standard addition methods applications. Infra-red spectroscopy. Sample preparation. Identification of compounds using correlation charts

#### Separation and Chromatographic Techniques

Separation and Chromatographic Techniques: Principles of adsorption, partition, ion exchange, molecular and affinity chromatography. Chromatogram description. Theoretical plates. Resolution selectivity. Gas-liquid chromatography. High performance liquid chromatography. Methods of separation. Mobile and stationary phases. Columns detectors solvents. Sample preparation. Qualitative and quantitative measurements. Use in the analysis of drugs. Internal standards. Ion chromatography. Solvent extraction

#### The Pharmaceutical industry

Introduction to pharmaceutical industry. Primary and secondary manufacture. Basic calculations in pharmaceutical analysis. Licensing of drugs and the pharmacopoeia. European Pharmacopoeia, BP and USP monographs of pure drugs and formulations including assays, storage and identification.

#### Pharmaceutical Chemistry

Selected categories of pharmaceuticals: vitamins, analgesics, general and local anaesthetics, sulphonamides antibiotics, antacids and acid-inhibitors, barbiturates.

#### Analytical Techniques

Analytical methods used in drug analysis – UV/Vis, AAS, AES, Titrimetric, IR, GC and HPLC methods. E(1%,1cm). Dissolution testing.

#### Chemistry of acids and bases

Ionisation of drugs. pKa values of drugs. Partition coefficients relating to drug absorption.

#### Extraction Methods in Pharmaceutical analysis

Excipients in formulations. Solvent and solid phase extraction methods

#### Control of the quality of analytical methods

Control of errors in analysis. Repeatability and reproducibility. Validation of drug analysis to include assay, precision specificity, LOD, LOQ, linearity, ruggedness and robustness.

#### Practical

Practicals will develop skills in the use of analytical equipment and techniques from AAS, AES, FTIR, HPLC, GC, and UV-Vis and ion chromatography. The practicals covered will reinforce and amplify the material covered in the lecture course. The use of updated software will be implemented.

Assessment Breakdown	%
Continuous Assessment	60.00%
Practical	40.00%

### Special Regulation

Students must achieve a minimum grade (35%) in both the practical and CA

### Continuous Assessment

Assessment Type	Assessment Description	Outcome addressed	% of total	Assessment Date
Examination	written assessment examination	1,5	20.00	n/a
Examination	written assessment examination.	1,5	20.00	n/a
Examination	written assessment examination	1,5	20.00	n/a

No Project

### Practical

Assessment Type	Assessment Description	Outcome addressed	% of total	Assessment Date
Practical/Skills Evaluation	Laboratory worksheets and reports Practical Log Book	2,3,4	40.00	n/a

No End of Module Formal Examination

**Module Workload**

<b>Workload: Full Time</b>		
<i>Workload Type</i>	<i>Frequency</i>	<i>Average Weekly Learner Workload</i>
Lecture	12 Weeks per Stage	4.00
Laboratory	12 Weeks per Stage	4.00
Estimated Learner Hours	15 Weeks per Stage	10.27
Total Hours		250.00

**Module Delivered In**

Programme Code	Programme	Semester	Delivery
CW_SABTP_B	<a href="#">Bachelor of Science (Honours) in Biosciences with Biopharmaceuticals</a>	4	Mandatory
CW_SABFQ_D	<a href="#">Bachelor of Science in Biosciences</a>	4	Mandatory
CW_SASCI_C	<a href="#">Higher Certificate in Science in Applied Biology or Applied Chemistry</a>	4	Group Elective 1