

RequirementsThis is prior learning (or a practical skill) that is mandatory before enrolment in this module is allowed.

No requirements listed

ZCHE C1101: Chemistry 1

University						
Module Title:			Chemistry 1			
Language of Instruction:		n:	English			
Credits: 5		5				
NFQ Level:		6				
Module Deli	vered In		6 programme(s)			
Teaching & Learning Strategies:			This module will be taught in two theory classes of 1 hour duration plus one 3 hour practical per week. To consolidate lectures and practicals, students will be required to carry out assignments and prepare a weekl practical report analysing their own research and results. Any course –related issue of questions that may arise will be discussed at lectures.			
Module Aim	:		To provide students with a knowledge of basic theoretical and practical chemistry principles.			
Learning Ou	ıtcomes					
On successfi	ul completion	n of th	nis module the learner should be able to:			
LO1	LO1 Describe basic chemical terminology, facts and principles.		chemical terminology, facts and principles.			
LO2	Compute stoichiometric, titrimetric, and other relevant calculations.					
LO3	Perform practical laboratory skills in chemical and analytical procedures		al laboratory skills in chemical and analytical procedures			
Pre-requisit	Pre-requisite learning					
Module Recommendations This is prior learning (or a practical skill) that is recommended before enrolment in this module.						
No recommendations listed						
Incompatible Modules These are modules which have learning outcomes that are too similar to the learning outcomes of this module.						
No incompatible modules listed						
Co-requisite Modules						
No Co-requis	No Co-requisite modules listed					

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Module Content & Assessment

Indicative Content

Atomic Theory

History & modern view of atomic structure. Electronic configuration, Pauli's exclusion principle, the Aufbau principle, and Hund's rule. Isotopes. Atomic Mass.

Periodic Table

Periodic table. How to read the table. Periodicity and trends, to include atomic radii, ionization energy, electron affinity, and electronegativity.

Bonding and Intermolecular Interactions

Molecules/lons/Metallic solids. Chemical bonds. Polarity. Van Der Waals Forces (London dispersion, dipole-dipole). Hydrogen bonding. States of matter - characteristics of solids, liquids, gases. Density.

Stoichiometry

Calculations with chemical formulae and equations: Formulae. Elemental analysis. Molecular mass. The mole. Chemical equations and stoichiometry.

Volumetric Analysis

Solutions. Concentrations: Molarity, ppm, percent solutions. Dilutions. Titrations: Apparatus, safety procedures, and technique. Titrimetric calculations.

Chemical Equilibrium

Dynamic equilibrium. Le Chatelier's principle to demonstrate effects on equilibrium. Applications in control of industrial processes. Haber process.

Acid base equilibria

Acids and bases - Bronsted-Lowry acids and bases. pH scale. Strong and weak acids. Buffer solutions.

Practicals

Use of volumetric glassware. Titrimetric analysis including direct, indirect, and back titrations using various acid-base examples. pH determination and equilibria reactions. Qualitative inorganic analysis.

Assessment Breakdown	%
Continuous Assessment	20.00%
Practical	50.00%
End of Module Formal Examination	30.00%

Special Regulation

Students must achieve a minimum grade (35%) in both the practical/CA and final examination

Continuous Assessment					
Assessment Type	Assessment Description	Outcome addressed	% of total	Assessment Date	
Examination	Throughout Year	1,2	20.00	n/a	

No Project

Practical				
Assessment Type	Assessment Description	Outcome addressed	% of total	Assessment Date
Practical/Skills Evaluation	Practical Log Book	3	50.00	Every Week

End of Module Formal Examination				
Assessment Type	Assessment Description	Outcome addressed	% of total	Assessment Date
Formal Exam	3 hour exam	1,2	30.00	End-of-Semester



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Module Workload

Workload: Full Time		
Workload Type	Frequency	Average Weekly Learner Workload
Lecture	12 Weeks per Stage	2.00
Practicals	12 Weeks per Stage	3.00
Independent Learning	15 Weeks per Stage	4.33
	Total Hours	125.00

Module Delivered In

Programme Code	Programme	Semester	Delivery
CW_SABTP_B	Bachelor of Science (Honours) in Biosciences with Biopharmaceuticals	1	Mandatory
CW_SABRE_B	Bachelor of Science (Honours) in Brewing and Distilling	1	Mandatory
CW_SAPHA_B	Bachelor of Science (Honours) in Pharmaceutics and Drug Formulation	1	Mandatory
CW_SAASC_D	Bachelor of Science in Analytical Science	1	Mandatory
CW_SABFQ_D	Bachelor of Science in Biosciences	1	Mandatory
CW_SASCI_C	Higher Certificate in Science in Applied Biology or Applied Chemistry	1	Mandatory