

Module Title:	Physical chemistry
Language of Instruction:	English
Credits:	5
NFQ Level:	7
Module Delivered In	1 programme(s)
Teaching & Learning Strategies:	To understand the importance of quantification and measurement in physicochemical process. To develop working models or rigs for efficient analyte extraction. To understand the importance of the use of instrumentation in industrial process.
Module Aim:	To provide a basic knowledge of the measurement techniques which underline chemical analysis and process. To use electrochemical and surface chemical methods to synthesize, extract and to electrodeposit. To investigate the mechanisms of chemical reaction.
Learning Outcomes	
<i>On successful completion of this module the learner should be able to:</i>	
LO1	Understand the use of common laboratory instrumentation in controlling physicochemical process.
LO2	Develop methods for the efficient production and monitoring of electrolyte in electrochemical process.
LO3	Understand the importance of efficiency and yield in processes.
Pre-requisite learning	
Module Recommendations <i>This is prior learning (or a practical skill) that is recommended before enrolment in this module.</i>	
No recommendations listed	
Incompatible Modules <i>These are modules which have learning outcomes that are too similar to the learning outcomes of this module.</i>	
No incompatible modules listed	
Co-requisite Modules	
No Co-requisite modules listed	
Requirements <i>This is prior learning (or a practical skill) that is mandatory before enrolment in this module is allowed.</i>	
No requirements listed	

Module Content & Assessment
Indicative Content

Websites and e-books will be used as much as formal printed material
n/a

Assessment Breakdown	%
Continuous Assessment	10.00%
Practical	30.00%
End of Module Formal Examination	60.00%

Special Regulation

Students must achieve a minimum grade (35%) in both the practical/CA and final examination.

Continuous Assessment

Assessment Type	Assessment Description	Outcome addressed	% of total	Assessment Date
Examination	Thermodynamics. Heats of reaction, entropy. Surface & Colloid Chemistry Viscosity and rheology. Electrochemistry, Kinetics Phase behaviour	1,2,3	10.00	n/a

No Project

Practical

Assessment Type	Assessment Description	Outcome addressed	% of total	Assessment Date
Practical/Skills Evaluation	Kinetic pathways of common reactions. Electrochemical extractions. Surface chemical methods and emulsions. Solutions and conductivity.	1,2,3	30.00	n/a

End of Module Formal Examination

Assessment Type	Assessment Description	Outcome addressed	% of total	Assessment Date
Formal Exam	Thermodynamics. Heats of reaction, entropy. Surface & Colloid Chemistry Viscosity and rheology. Electrochemistry, Kinetics Phase behaviour	1,2,3	60.00	End-of-Semester

SETU Carlow Campus reserves the right to alter the nature and timings of assessment

Module Workload

Workload: Full Time		
<i>Workload Type</i>	<i>Frequency</i>	<i>Average Weekly Learner Workload</i>
Lecture	30 Weeks per Stage	1.00
Laboratory	30 Weeks per Stage	1.00
Total Hours		60.00

Module Delivered In

Programme Code	Programme	Semester	Delivery
CW_SASES_B	Bachelor of Science (Honours) in Environmental Science	3	Mandatory